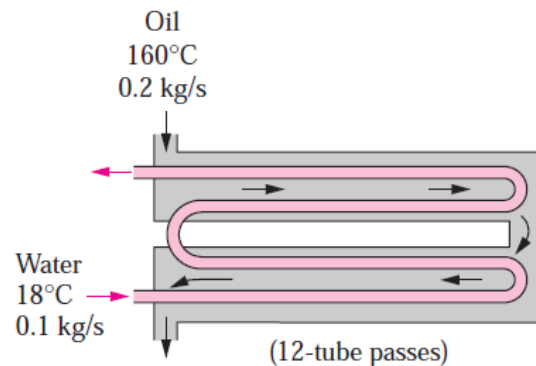


# Mech 262- Thermodynamics 2

## Assignment 7

### Question 1

Hot oil ( $C_p = 2200 \text{ J/kg} \cdot ^\circ\text{C}$ ) is to be cooled by water ( $C_p = 4180 \text{ J/kg} \cdot ^\circ\text{C}$ ) in a 2-shell-passes and 12-tube-passes heat exchanger. The tubes are thin-walled and are made of copper with a diameter of 1.8 cm. The length of each tube pass in the heat exchanger is 3 m, and the overall heat transfer coefficient is  $340 \text{ W/m}^2 \cdot ^\circ\text{C}$ . Water flows through the tubes at a total rate of  $0.1 \text{ kg/s}$ , and the oil through the shell at a rate of  $0.2 \text{ kg/s}$ . The water and the oil enter at temperatures  $18^\circ\text{C}$  and  $160^\circ\text{C}$ , respectively.



Determine the rate of heat transfer in the heat exchanger and the outlet temperatures of the water and the oil. (Answers:  $36.2 \text{ kW}$ ,  $104.6^\circ\text{C}$ ,  $77.7^\circ\text{C}$ )

### Question 2

Consider an aluminum cold drink can that is initially at a uniform temperature of  $3^\circ\text{C}$ . The can is  $12.5 \text{ cm}$  high and has a diameter of  $6 \text{ cm}$ . If the convection heat transfer coefficient between the can and the surrounding air at  $25^\circ\text{C}$  is  $10 \text{ W/(m}^2 \cdot ^\circ\text{C)}$ , determine how long it will take for the average temperature of the drink to rise to  $10^\circ\text{C}$ .

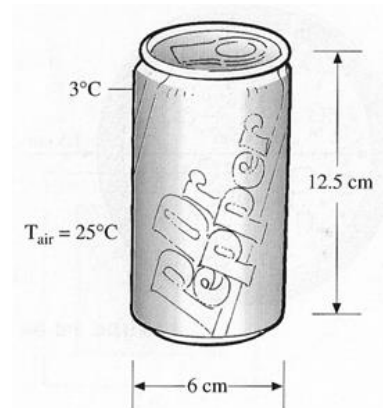


FIGURE P8-98

In an effort to slow down the warming of the cold drink a person puts the can in a perfectly fitting  $1 \text{ cm}$  thick cylindrical rubber insulation [ $k = 0.13 \text{ W/(m} \cdot ^\circ\text{C)}$ ]. Now how long will it take for the average temperature of the drink to rise to  $10^\circ\text{C}$ ? Assume the top of the can is not covered.

### Question 3

In a manufacturing facility,  $2 \text{ in}$  diameter brass balls ( $k = 64.1 \text{ Btu/h} \cdot \text{ft} \cdot ^\circ\text{F}$ ,  $\rho = 532 \text{ lbm/ft}^3$ , and  $C_p = 0.092 \text{ Btu/lbm} \cdot ^\circ\text{F}$ ) initially at  $250^\circ\text{F}$  are quenched in a water bath at  $120^\circ\text{F}$  for a period of  $2 \text{ min}$  at a rate of  $120 \text{ balls per minute}$ .

If the convection heat transfer coefficient is  $42 \text{ Btu/h} \cdot \text{ft}^2 \cdot ^\circ\text{F}$ , do the following:

- Convert this problem to SI units.
- Determine the temperature of the balls after quenching.
- Determine the rate at which heat needs to be removed from the water in order to keep its temperature constant at  $120^\circ\text{F}$ .

